# NITRATE CONCENTRATION IN GROUNDWATER: A CROSS-SECTIONAL STUDY IN THREE VILLAGES NEAR PADDY FIELDS IN BACHOK DISTRICT, KELANTAN DURING THE HARVEST SEASON

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#### ABSTRACT

**Introduction:** Nitrate (NO<sub>3</sub><sup>-</sup>) is essential for producing healthy plants. However, excessive nitrate not taken up may eventually accumulate in groundwater, which is detrimental to human health if consumed. The maximum acceptable concentration for nitrate in drinking water is 44.3 ppm. **Methodology:** Keting, Kuchelong and Telaga Ara villages in Bachok district, Kelantan were chosen because of their proximity to paddy fields. Houses using groundwater for drinking and cooking purposes were selected and other data such as age and depth of wells, and distance to nitrate source were recorded. Data were collected during the harvest season of October and November of 2018 and two replicates of water samples were analysed using a portable meter with an attached nitrate electrode. **Result:** A total of 136 wells were selected. Nitrate concentration in Kuchelong village ranged from 0.09 to 8.12 ppm (mean 0.74 ± SD 1.82 ppm), in Keting village ranged from 0.81 to 28.8 ppm (mean 3.76 ± SD 5.49 ppm) and in Telaga Ara village ranged from 0.95 to 43.0 ppm (mean 8.34 ± SD 7.70 ppm). Nitrate readings did not exceed the maximum acceptable concentration. One-way ANOVA analysis shows that there is a significant difference in nitrate concentration between villages (p<0.05), while Spearman's rho correlation analysis shows that there are no significant relationships between nitrate concentration and variables (p>0.05). **Conclusion:** Water samples have nitrate with concentrations below the maximum acceptable level. Periodical analysis of groundwater should be performed to ensure nitrate remain below the maximum acceptable level.

Keywords: Nitrate, Well water, Bachok, paddy, lower

### 1. Introduction

Nitrogen gas  $(N_2)$  is the most abundant element in the atmosphere (78%), but it cannot be readily used by plants for growth purposes. However, by the process of fixing and changing atmospheric  $N_2$  into organic nitrogen (N) forms by certain symbiotic and non-symbiotic organisms in the soil, these can then be transformed into inorganic ammonium (NH<sub>4</sub><sup>+</sup>) in the ammonification process, which plants can utilize (USDA, 1995). Organic N is naturally occurring but can also be added into the soil as plant residues and manure.

Nitrate (NO<sub>3</sub>) is a polyion made up from one nitrogen and three oxygen atoms and it is extensively

used in agricultural activities as fertilizers. In the natural world, nitrate, which is a component of the Nitrogen Cycle, is synthesized from ammonia (NH<sub>3</sub>) in the soil by nitrifying aerobic bacteria, including from the genera *Nitrosomonas* which converts ammonia to nitrites (NO<sub>2</sub><sup>-</sup>), and *Nitrobacter*, which then converts nitrites to nitrate (Encyclopaedia Brittanica, 2019). Nitrate-rich soils can be found in many parts of the world, including deserts (Erickson, 1981). Man-made sources of nitrate include septic tanks, fertilizers and wastewater treatment (WHO, 2011).

Groundwater is important for residents especially in the rural areas in order to fulfill their daily needs. It can be contaminated by nitrate originating from human activities such as agriculture, industries and disposal of solid waste, among others (Aida Soraya et al., 2014). In China, a study by Ju et al. (2006) found that extremely high fertilizer N inputs has resulted in large amounts of nitrate accumulated in the vegetable soils and the shallow groundwater was heavily contaminated by nitrate-N. In addition, N is more easily lost to streams and groundwater from ecosystems than any other essential elements (Vitousek et al., 2002).

The presence of nitrate in groundwater can pose a health risk to residents. Parvizishad et al. (2016) stated that consumption of drinking water and food containing high nitrites and nitrates may cause diseases such as cancer, methemoglobinemia, enlargement of the thyroid gland and diabetes mellitus. However, nitrate may only pose a risk to human health when its concentration exceeds the maximum acceptable value of nitrate in drinking water. In Malaysia, the maximum acceptable value for nitrate in drinking water is 10mg/L NO<sub>3</sub>-N (Engineering Services Division, Minitry of Health, Malaysia, 2010).

According to the United Sates Environmental Protection Agency (USEPA), among the water quality problems associated with nitrification are nitrite/nitrate formation, dissolved oxygen depletion and increases in heterotrophic plate count (HPC), ammonia oxidizing bacteria (AOB) and nitrite oxidizing bacteria (NOB) (USEPA, 2002).

About 70% of residents in Kelantan still rely on groundwater as their main source of drinking water (Aida Soraya et al., 2016) and this indicates the important water safety aspect in the context of soluble chemicals that can have health effects on residents. Nas and Berktay (2006) stated that approximately 75% of the water supply in the city of Konya in Turkey relies on groundwater, which is supplied by 198 wells.

Most houses in Bachok district in Kelantan state are located near paddy fields (Muhammad Nur Fakhri and Shaharuddin, 2017) and agricultural activities are the main occupation for residents in the area (Aida Soraya et al., 2016).

The aims of this study were to determine nitrate concentration in groundwater during the harvest season as usage of fertilizer (source of nitrate) in Bachok district, Kelantan, to determine any significant difference in nitrate concentration between villages, and to determine any significant relationships in nitrate concentration with age and depth of wells, and distance from nitrate source. The problem to be highlighted here is whether nitrate concentration in groundwater in the study areas was safe for drinking purposes as the residents rely on groundwater as their main water source.

## 2. Materials and Method

### 2.1 Description of study area

Three villages were chosen for this study, namely Kuchelong, Keting and Telaga Ara villages. (Figure 1).

#### 2.2. Well water sampling and analysis

A Hanna Instrument HI98191 Professional Waterproof Portable ph/ORP/ISE Meter with a HI4113 Nitrate Combination Ion Selective Electrode (ISE) was used to analyze the water samples. Samples were collected in three replicates and were then stored in high density polyethylene (HDPE) bottles before analysis.

### 2.3. Statistical analysis

The collected data were analyzed using IBM SPSS (Statistical Package for Social Science) version 22. The distribution of variables in this study was analyzed using descriptive statistic including standard deviation, mean and median.

### 3. Results and discussions

This study was conducted from 27 October 2018 until 02 November 2018. At the time, it was the paddy harvest season in Bachok district. A total of 136 wells were analysed for nitrate from three villages in Bachok district. The villages were Kuchelong with 43 wells, Keting with 36 wells and Telaga Ara with 57 wells (Table 1). These villages were located approximately 20 kilometres southeast of Kota Bharu, the capital of Kelantan state. Table 1.2 highlight the age, depth and distance of the well from nitrate resources.

Table 1: Number of wells according to villages

| Village    | No. of wells |  |  |  |
|------------|--------------|--|--|--|
| Kuchelong  | 43           |  |  |  |
| Keting     | 36           |  |  |  |
| Telaga Ara | 57           |  |  |  |
|            |              |  |  |  |

N=136



Figure 1: Map showing location of study area. Inset: Map showing Bachok district within Kelantan state.

Nitrate concentration in Kuchelong village ranged from 0.09 to 8.12 ppm with a mean of 0.74  $\pm$  SD 1.82 ppm, while in Keting village, it ranged from 0.81 to 28.8 ppm with a mean of 3.76  $\pm$  SD 5.49 ppm. In Telaga Ara village, nitrate concentration ranged from 0.95 to 43.0 ppm with a mean of 8.34  $\pm$  SD 7.70 ppm (Table 2 and Figure 2). From 136 wells, none of the wells had readings above the recommended level (44 ppm) by the Malaysian health authority. It was found that all readings from the three villages were below the maximum acceptable level of 44.3 ppm NO<sub>3</sub><sup>-</sup> as stated by WHO (Table 3).

In this context, the water sampling period was conducted during the harvest season in the study area. As such, no more fertilization of soil is occurring as it is not needed for plant growth and therefore, concentration of nitrate in soil and water adjacent to the paddy fields are lower. In a case study conducted at lower Mae Klong river basin in Thailand, Jaturong, et al. (2015) found that from 73 wells sampled for nitrate, only 4 samples exceeded the acceptable level of groundwater quality for drinking purposes. The authors also mentioned that land-use patterns played an important role in nitrate contamination. Where there were agricultural areas such as sugarcane planting areas and mixed orchard area, the concentration detected was more than the acceptable level. This can also be said for this study, where a number of wells where concentration of nitrate was higher than the mean were located near sources of nitrate contamination.

One-way ANOVA was used to compare nitrate concentrations between villages. Analysis showed that there is a significant difference between all three villages, where p<0.05 (Table 4). Observations on sampling sites with readings higher than the mean which shows that the wells in question were near a source of nitrate. For example, one well in Keting village with a nitrate concentration of 28.8 ppm was located near the paddy field and animal farm.

| Village         |       | Keting  | Kuchelong | Telaga<br>Ara |
|-----------------|-------|---------|-----------|---------------|
| No. of<br>wells | N     | 36      | 43        | 57            |
| Age             | <5    | 4       | 15        | 10            |
|                 | years |         |           |               |
|                 | 6-10  | 7       | 4         | 5             |
|                 | years |         |           |               |
|                 | 11-15 | 1       | 3         | 3             |
|                 | years |         |           |               |
|                 | 16-20 | 9       | 5         | 7             |
|                 | years |         |           |               |
|                 | >20   | 15      | 16        | 32            |
|                 | years |         |           |               |
| Depth of        | <5    | 0       | 3         | 0             |
| well            | meter |         |           |               |
|                 | 5-10  | 18      | 29        | 23            |
|                 | meter |         |           |               |
|                 | 11-15 | 15      | 9         | 22            |
|                 | meter |         |           |               |
|                 | >15   | 3       | 2         | 12            |
|                 | meter |         |           |               |
| Distance        | Mean  | 158.38± | 483.72±   | 212.84±       |
| from            |       | 111.31  | 297       | 271.212       |
| source          |       |         |           |               |
| (meter)         |       |         |           |               |

| Table 1.2 : Age, | depth | and | distance | of well from |
|------------------|-------|-----|----------|--------------|
| nitrate source.  |       |     |          |              |

 Table 2: Mean and range of nitrate concentration in water samples from the three villages (N= 136)

| Mean (ppm)            | Range (ppm)                                    |
|-----------------------|--|
| 0.74 <u>+</u> SD 1.82 | 0.09 – 8.12                                    |
| 3.76 <u>+</u> SD 5.49 | 0.81 – 28.8                                    |
| 8.34 <u>+</u> SD 7.70 | 0.95 – 43.0                                    |
|                       | 0.74 <u>+</u> SD 1.82<br>3.76 <u>+</u> SD 5.49 |



Figure 2: Concentration of nitrate based on villages

 Table 4: One-way ANOVA for comparison of nitrate concentration between villages

|           |      |      | 95% Confidence    |        | Sig.   |
|-----------|------|------|-------------------|--------|--------|
|           |      |      | Interval for Mean |        |        |
|           |      |      |                   |        | _      |
|           |      |      | Lower             | Upper  |        |
| Village   | Mean | SD   | Bound             | Bound  |        |
| Keting    |      |      |                   |        | <0.001 |
| (         | 3.76 | 5.49 | 1.8981            | 5.6147 |        |
| (n = 36)  |      |      |                   |        |        |
| Kuchelong |      |      |                   |        | _      |
|           | 0.72 | 1.82 | .1628             | 1.2856 |        |
| (n = 43)  |      |      |                   |        |        |
| Telagara  |      |      |                   |        | -      |
|           | 8.34 | 7.70 | 6.2985            | 10.382 |        |
| (n = 57)  |      |      |                   |        |        |
|           |      |      |                   |        |        |

Another example, a well in Telaga Ara village with a nitrate concentration of 43.3 ppm, was also located near the paddy field and animal farm. Sahoo, Kim and Powell (2016) has stated that all livestock operations should be located at least 100 - 200 ft (30.5 - 61 m) away from any wells to reduce the risk of water pollution. Spearman's rho correlation test was conducted to see any significant relationship in nitrate concentration with three variables, namely age and depth of well and distance from nitrate source (Table 5).

**Table 5:** Correlation test between nitrateconcentration with three variables i.e. age, depth anddistance to nitrate resources

|                    |      | Age  | Depth | Distance |
|--------------------|------|------|-------|----------|
|                    |      |      |       | From     |
|                    |      |      |       | Nitrate  |
|                    |      |      |       | Source   |
|                    |      |      |       |          |
| Keting village     | r    | .192 | .082  | .165     |
|                    |      |      |       |          |
|                    | Sig. | .262 | .634  | .336     |
| Kuchelong village  | r    | 122  | 206   | .152     |
|                    | Sig. | .435 | .186  | .331     |
| Telaga Ara village | r    | .031 | .191  | .022     |
|                    | Sig. | .822 | .155  | .873     |

Note: Spearman's rho

From the analysis, it was found that there are no significant relationships with all three variables. Age and depth of well and distance from nitrate source did not play a role in nitrate concentration in the wells studied. These results were different from a study done by Swistock, Clemens and Sharpe (2009) who stated that nitrate concentration was correlated with the distance to the nearest cornfield and crop fields. Also, from the same study, the authors found that age of the well was statistically important, while depth of well is modestly significant.

## 4. Conclusion

In conclusion, all groundwater samples have nitrate concentration lower than maximum acceptable level. One reason behind this is that the sampling period is conducted during the harvest season. Recommendations should be made to residents with higher concentration of nitrate in their drinking water on ways to decrease nitrate to an acceptable level can reduce the health effects of nitrate pollution in the long term. Awareness on water quality problems should be instilled into residents by way of education in order for them to be resilient. In conclusion, all groundwater samples have nitrate concentration lower than maximum acceptable level. One reason behind this is that the sampling period is conducted during the harvest season. Recommendations should be made to residents with higher concentration of nitrate in their drinking water on ways to decrease nitrate to an acceptable level can reduce the health effects of nitrate pollution in the long term. Awareness on water quality problems should be instilled into residents by way of education in order for them to be resilient.

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